

# Integrating Solid-State Hydrogen Storage Materials into Town Planning for a Sustainable Urban Future

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**Organization:** Technological and Higher Education Institute of  
Hong Kong (THEi)



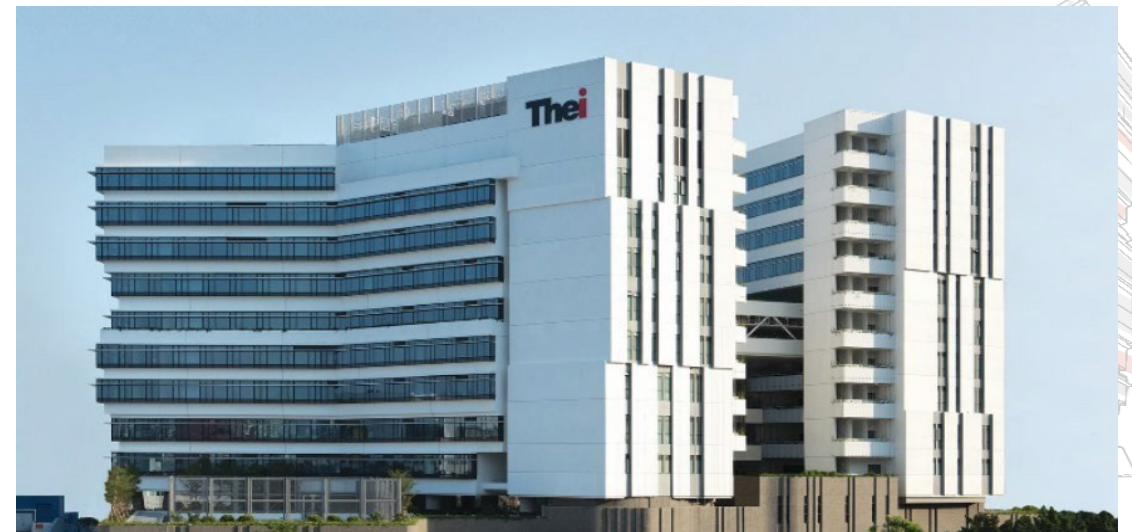
# THEi

Ir Dr. Alex Tsang (MAT, CML)

*FIChemE, FIMMM, FIET, FHEA*

Programme Leader of BSc in Green  
Engineering & Sustainability

Research interest: hydrogen energy, biomass energy,  
battery technology, green chemical production,  
techno-economic analysis



# Carbon Neutral Initiatives

- *Hong Kong Government initiative on Carbon Reduction:*

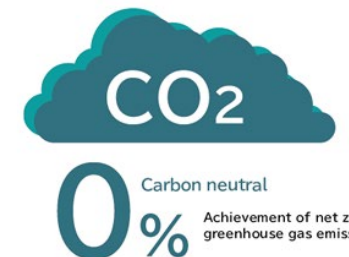
Hong Kong plays a part to help fulfill the obligations that China has under the Paris Agreement. As such, Hong Kong will need to review our **climate change efforts** every 5 years and align them with the submission timelines under the Paris Agreement.

- *Hong Kong's 2030 Target: Carbon Peak*

Hong Kong will **reduce its carbon intensity** by 65% to 70% using 2005 as the base. (*Hong Kong's Climate Action Plan 2030+*)



Carbon neutral  
by 2050







0% Carbon neutral  
Achievement of net zero  
greenhouse gas emission

# Green Hydrogen

There are many 'colors' of hydrogen – each referring to how it is produced.

**Green hydrogen** is the only variety produced in a climate-neutral manner.

It could play a vital role in global efforts to reach net-zero emissions by 2050.

Color	GREY HYDROGEN	BLUE HYDROGEN	TURQUOISE HYDROGEN*	GREEN HYDROGEN
Process	SMR or gasification  96% (IRENA)	SMR or gasification with carbon capture (85-95%)	Pyrolysis	Electrolysis
Source	Methane or coal  	Methane or coal  	Methane  	Renewable electricity  
	1.5–2.5 USD/kg			2.5–7 USD/kg

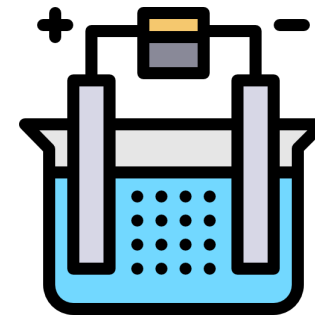


# Hydrogen Supply in Hong Kong

- Towngas

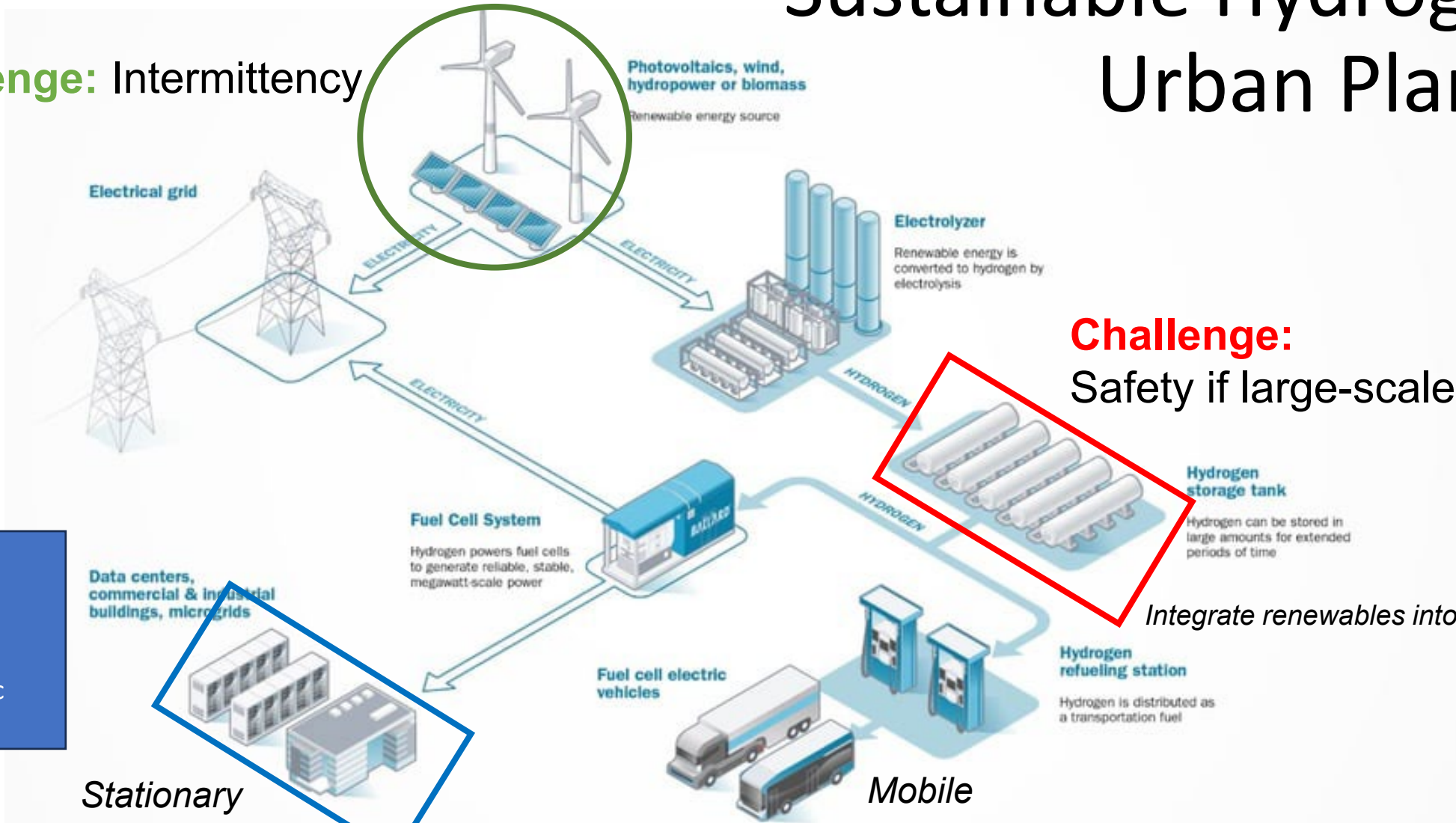


- Other H<sub>2</sub> suppliers, e.g. Linde



# Sustainable Hydrogen in Urban Planning

**Challenge:** Intermittency



**Challenge:** Safety if large-scale storage

*Integrate renewables into various sectors*

Large scale:  
 ≥ metric tons

Small scale:  
 kg to ≤ metric tons

Stationary

Mobile



# Sustainable Hydrogen in Urban Planning

The successful development of hydrogen storage solutions is essential for the penetration of hydrogen at each level of the energy supply chain, e.g. backup power, data centers, hospitals, etc.



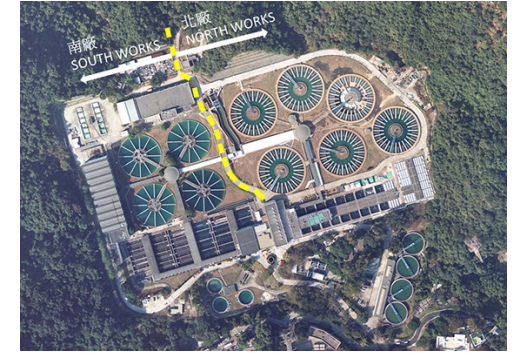
Residential area



Data Centre Power Outage  
*Image:* <https://powerwhips.com>



Hospital blackouts: rising death toll  
*Image:* NBC News



Water treatment plants  
[www.wsd.gov.hk](http://www.wsd.gov.hk)



Airport – runway lighting  
[www.stantec.com](http://www.stantec.com)



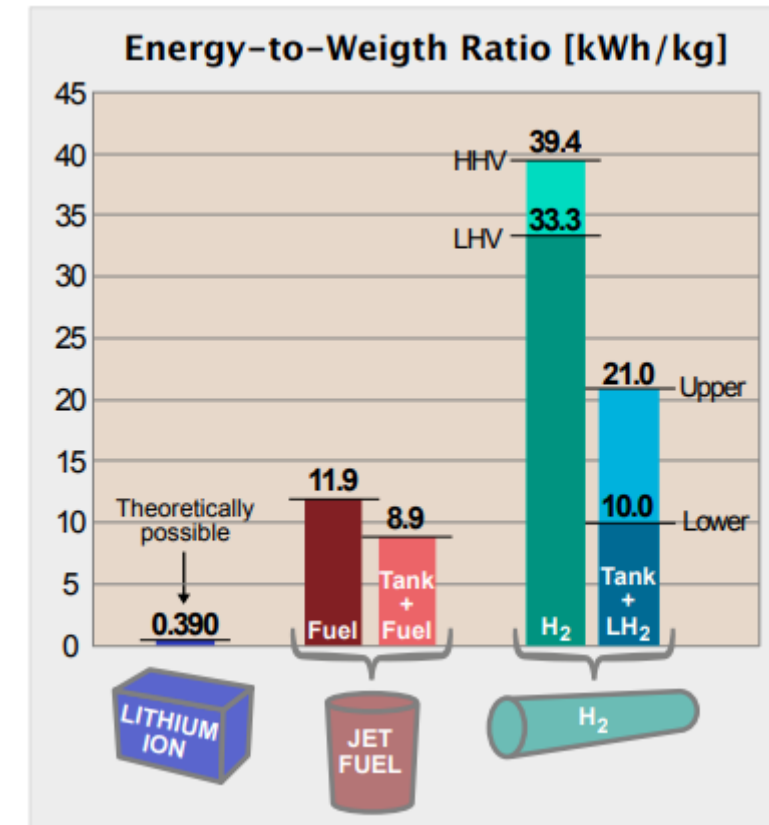
Telecommunication facilities  
[www.securitysales.com/](http://www.securitysales.com/)



Emergency operations center  
[avispl.com/](http://avispl.com/)

# Large-scale H<sub>2</sub> storage vs battery

1. higher energy density – H<sub>2</sub> (39.4 kWh/kg) vs battery (0.15 - 0.39 kWh/kg) more energy per unit volume or mass → less space
2. Scalability - H<sub>2</sub> relatively straightforward , battery needs adding more individual battery units → more complex and costly
3. Long-duration storage - H<sub>2</sub> extended periods without significant losses, battery energy loss due to chemical reactions, leakage current and internal resistance, etc



Source: IEEE Electrification Magazine 2021, 9, 92-102.



# Current Technology

- **Type I:** withstand only up to 50 MPa
- **Type II:** 30-40% lighter, steel liner
- **Type III:** liner Al, glass fiber
- **Type IV:** plastic liner wrapped with carbon fiber and other composite materials, lightweight, durable, and have high storage capacity (H<sub>2</sub> density: 5.7 wt%, 70 MPa, 40 kg/m<sup>3</sup>)

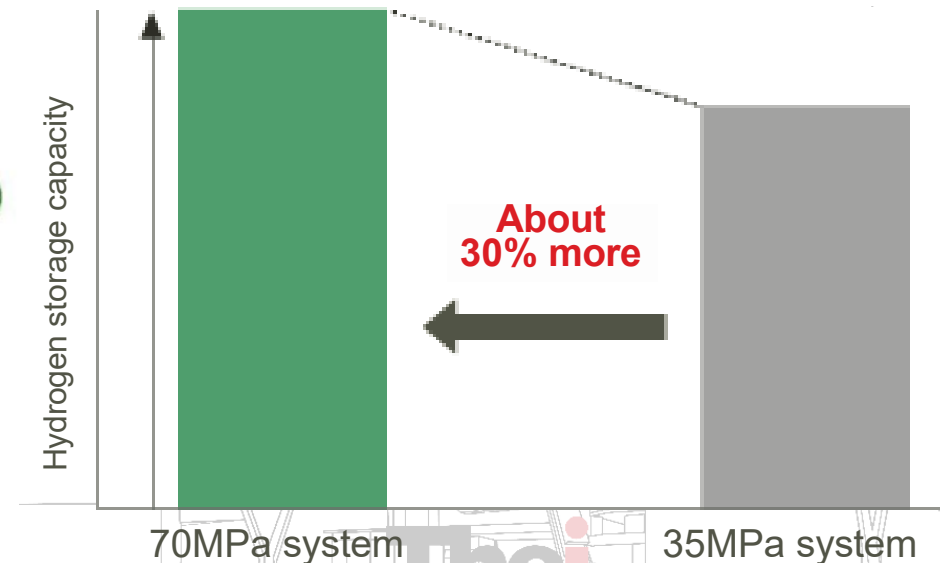
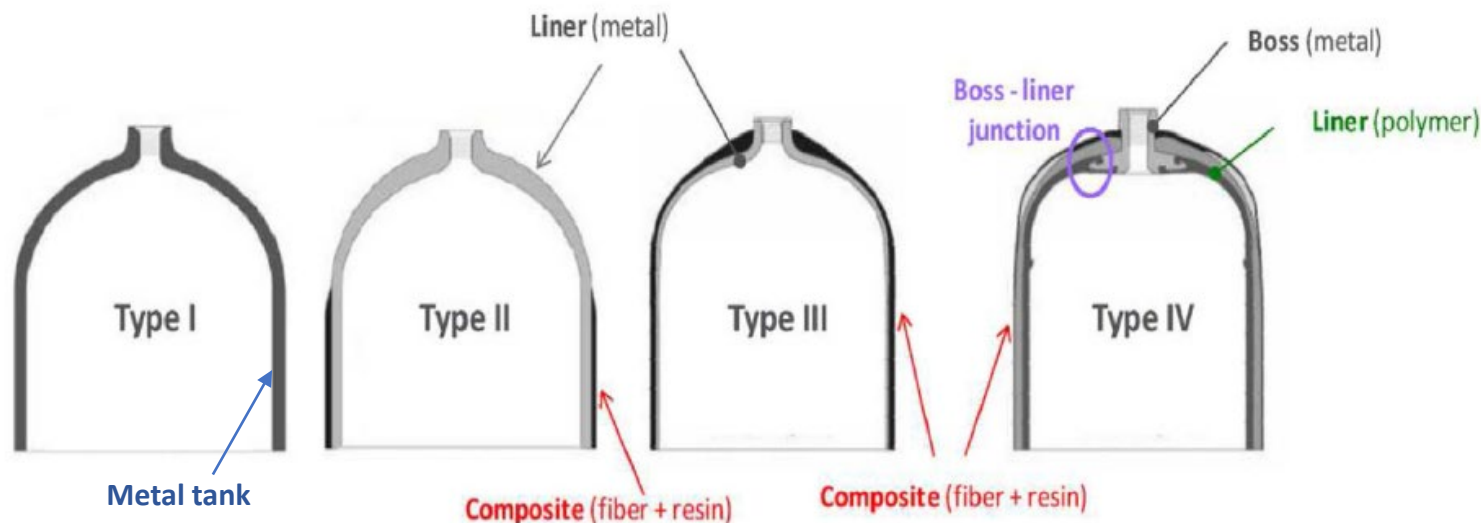
Weight ↔ Cost



*Toyota Mirai*

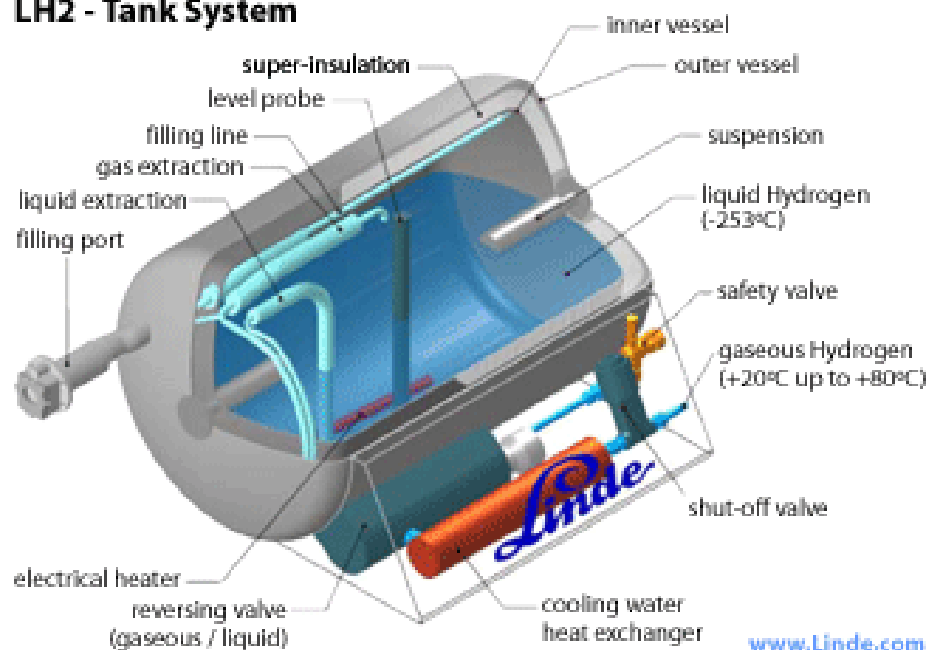


3.1 kW/L



# Current Technology

## LH2 - Tank System

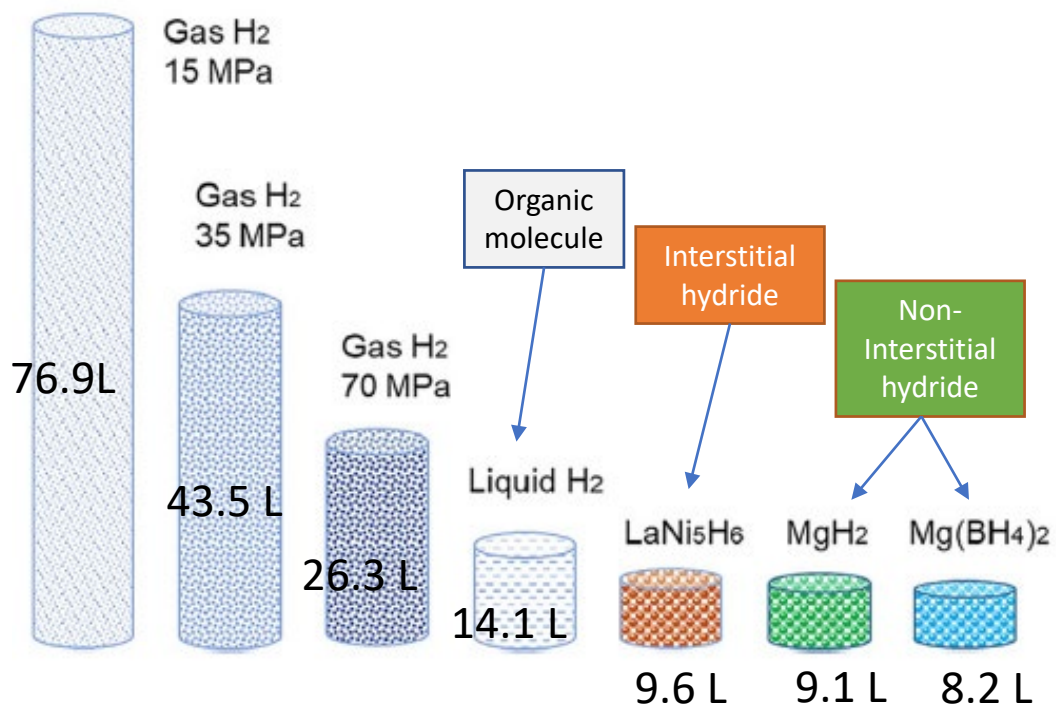


- High H<sub>2</sub> density: 70kg/m<sup>3</sup>
- Low temperature (-250°C)
- Consume up to 35% of the energy in the stored H<sub>2</sub>
- boil-off losses
- Larger scale → less boil-off losses ( $\propto$  surface area to volume)



BMW Hydrogen 7

# H<sub>2</sub> Storage Materials (HSM) (Potential for Large-Scale Storage)



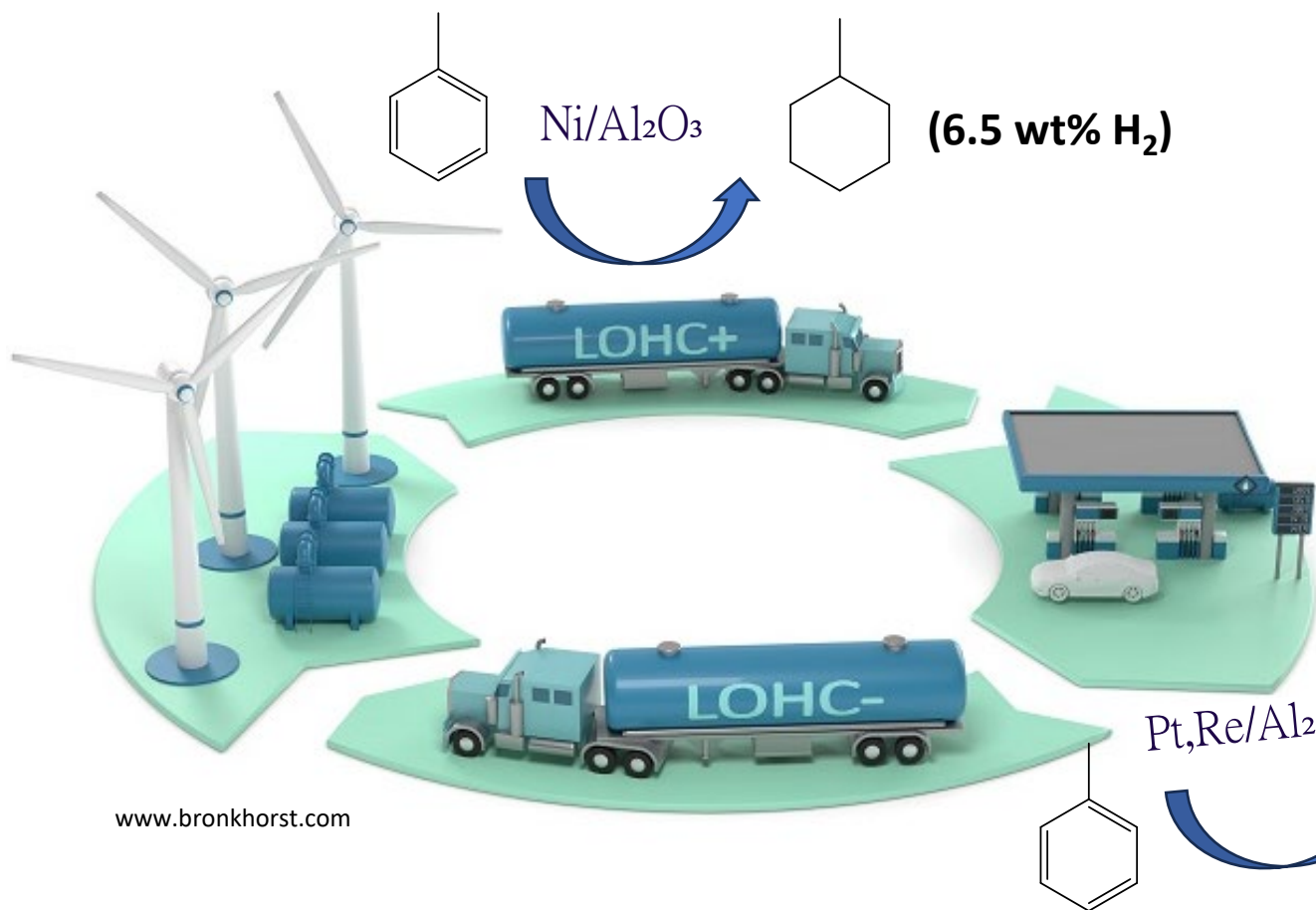
Comparison on the volumes for 1 kg H<sub>2</sub> (11,200 L) in various methods and materials.

*Advantages of Solid-State HSM:* High H<sub>2</sub> storage capacity, safe, stability, portable devices and transportation

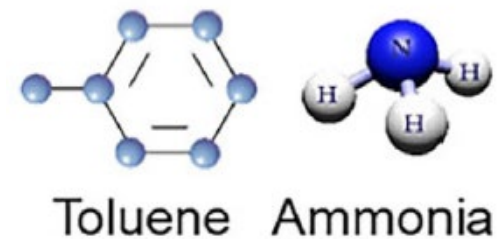
*Disadvantage:* Limited kinetics, require high temp/pressure, costly, cycling problem



# Liquid organic hydrogen carrier (LOHC)

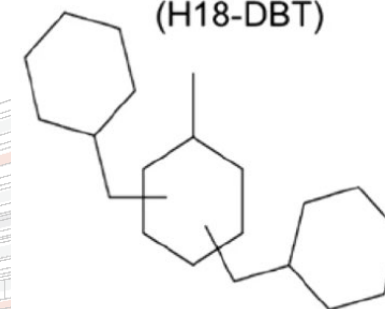


www.bronkhorst.com



Formic acid (CH<sub>2</sub>O<sub>2</sub>)  
(4.4 wt% H<sub>2</sub>)

perhydro  
dibenzyltoluene  
(H18-DBT)



6.2 mass% H<sub>2</sub>

Organic Liquid

(e.g. cycloalkanes, heterocycles, ammonia)  
Accounts of Chemical Research 2017, 50, 74.

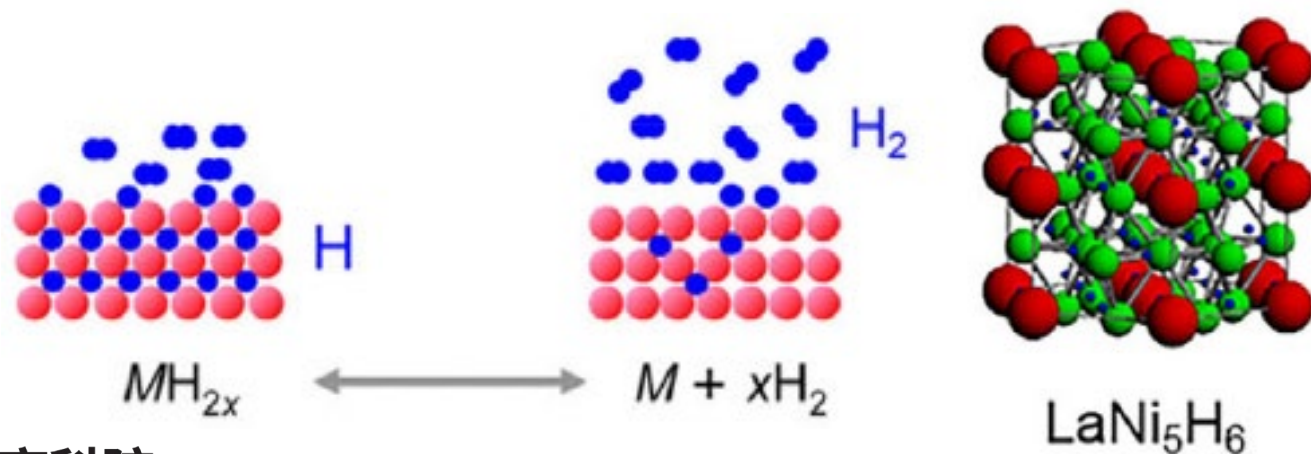
www.axens.net

# LaNi<sub>5</sub>H<sub>6</sub> (interstitial)

- H<sub>2</sub> forms metallic bonds in Interstitial hydride Materials

*Advantages:* Reversibility, safety

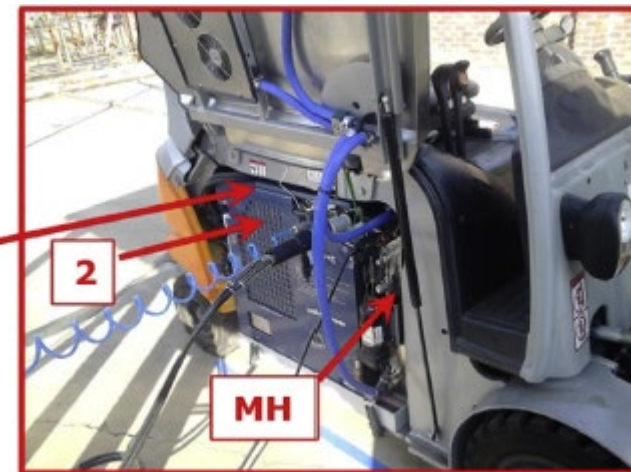
*Disadvantage:* Slow kinetics, materials degradation



AB<sub>5</sub>-type alloy (1.88 wt% H<sub>2</sub>)

*Forklift application*

1. DC/AC
2. Filling the MH



3. Compressor and Dispensing system

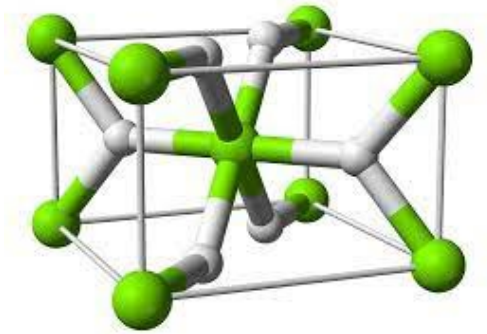


HySA Systems Competence Centre

# MgH<sub>2</sub> (non-interstitial)

*Advantages:* Abundant and low-cost, Reversibility

*Disadvantage:* Slow kinetics, materials degradation, high temperature requirement



MgH<sub>2</sub> – Rutile structure (H/M = 2)



MgH<sub>2</sub> (7.6 wt% H<sub>2</sub>)  
POWERPASTE

Direct hydrogenation of Mg metal at high pressure and temperature (200 atmospheres, 500 °C)

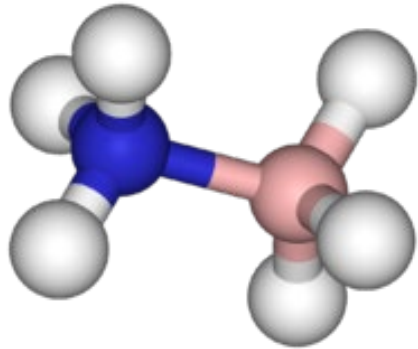
At 287 °C it decomposes to produce H<sub>2</sub> at 1 bar pressure. The high temperature required is seen as a limitation in the use of MgH<sub>2</sub> as a reversible hydrogen storage medium.



MgH<sub>2</sub> also readily reacts with water to form hydrogen gas



# Solid Borane Materials

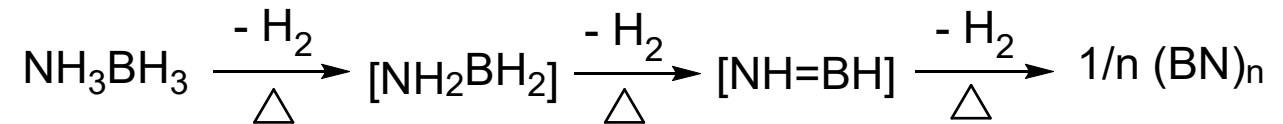


Ammonia borane (NH<sub>3</sub>BH<sub>3</sub>)  
(19.6 wt%)

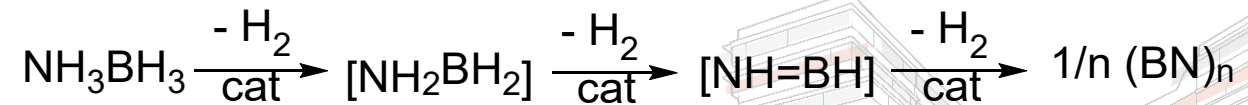


*Advantages:* Fast kinetics, high H<sub>2</sub> wt%  
*Disadvantage:* Reversibility, cycling problem

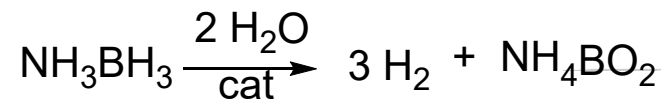
- Thermolysis



- Catalytic dehydrocoupling

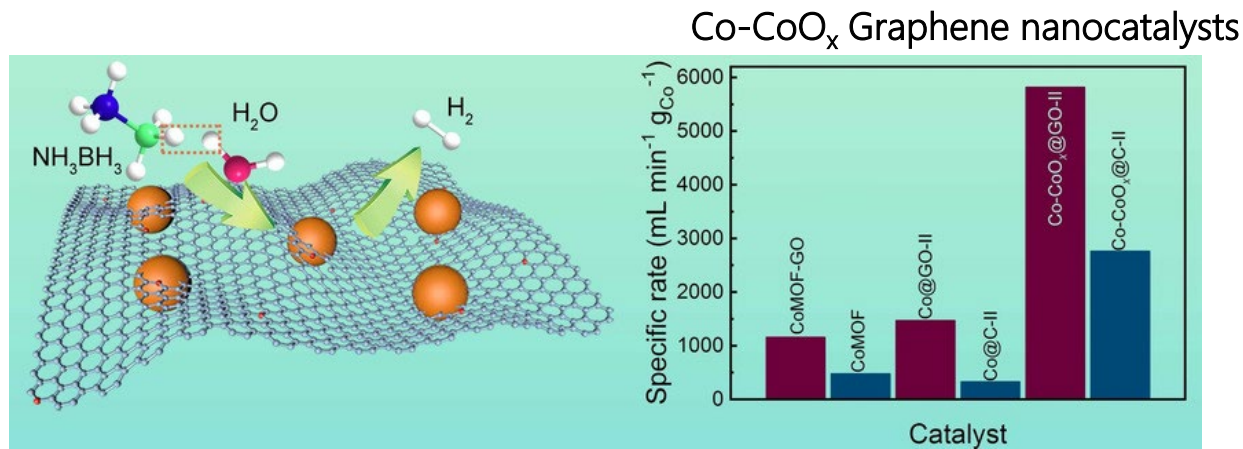


- Catalytic hydrolysis

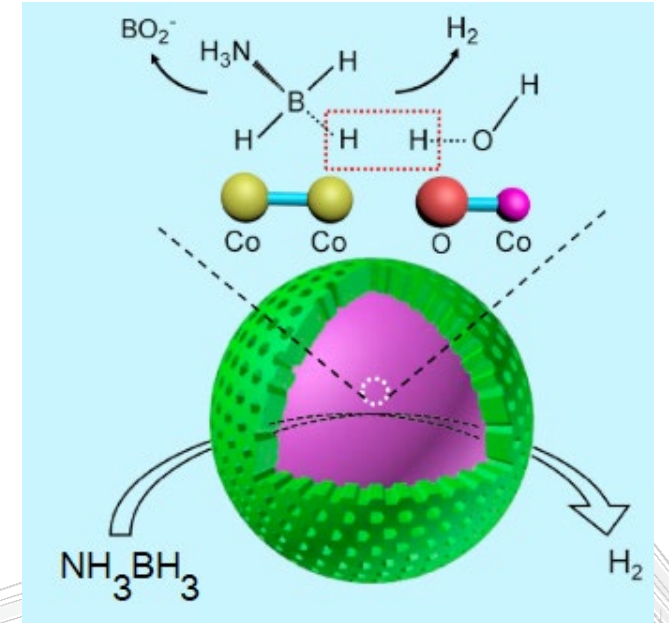


Require an efficient catalyst!

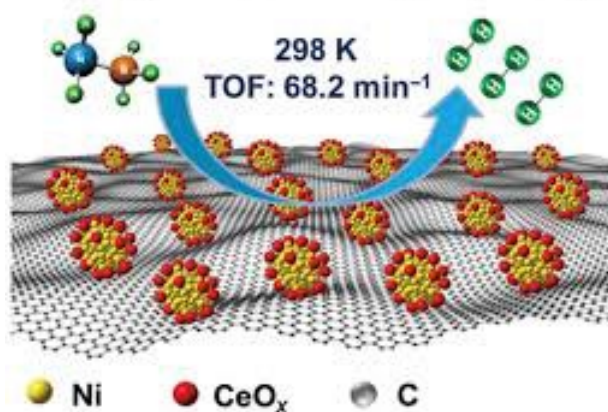
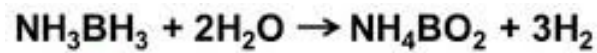
# Solid Borane Materials Research



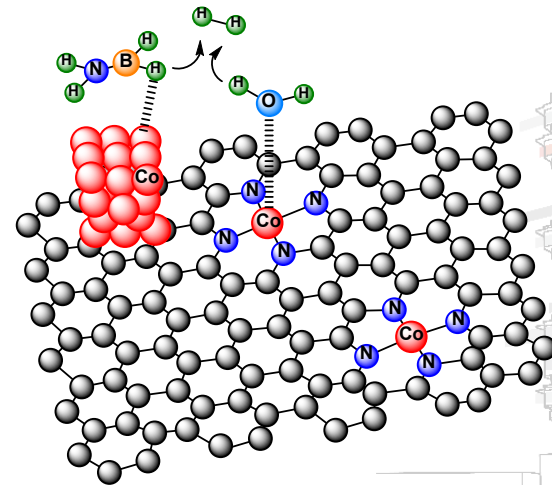
Chinese – An Asian Journal, 2020, 15, 3087-309.



ACS Sustainable Chemistry & Engineering 2019, 7, 9782-9792.



Nano Research 2018, 11, 4412-4422.



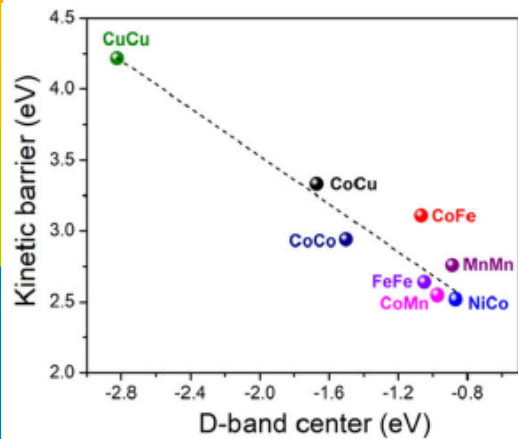
Journal of Materials Chemistry A 2021, 10, 5580-5592.

# Our Research

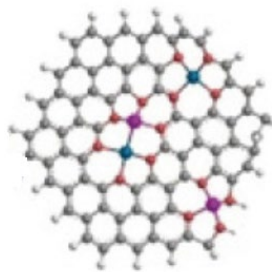
- To use single atom strategy and past experience to solve challenging energy and emerging environmental problem.

## Research & Catalyst Development

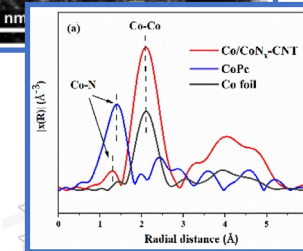
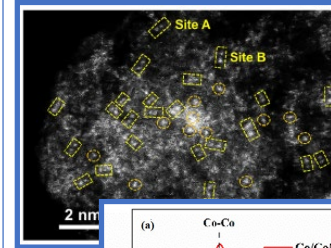
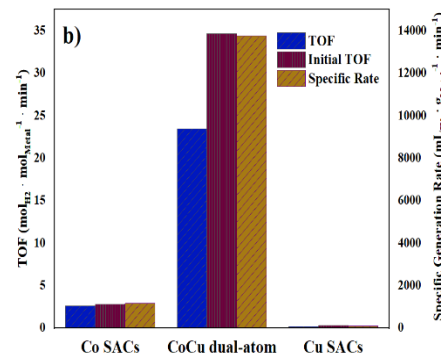
### Theoretical Screening



Experimental Design:  
Support the dual-atom  
on carbon support

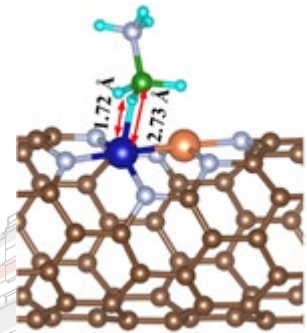


### Catalytic Evaluation



Characterization  
e.g. HAADF-STEM,  
XANES and EXAFS

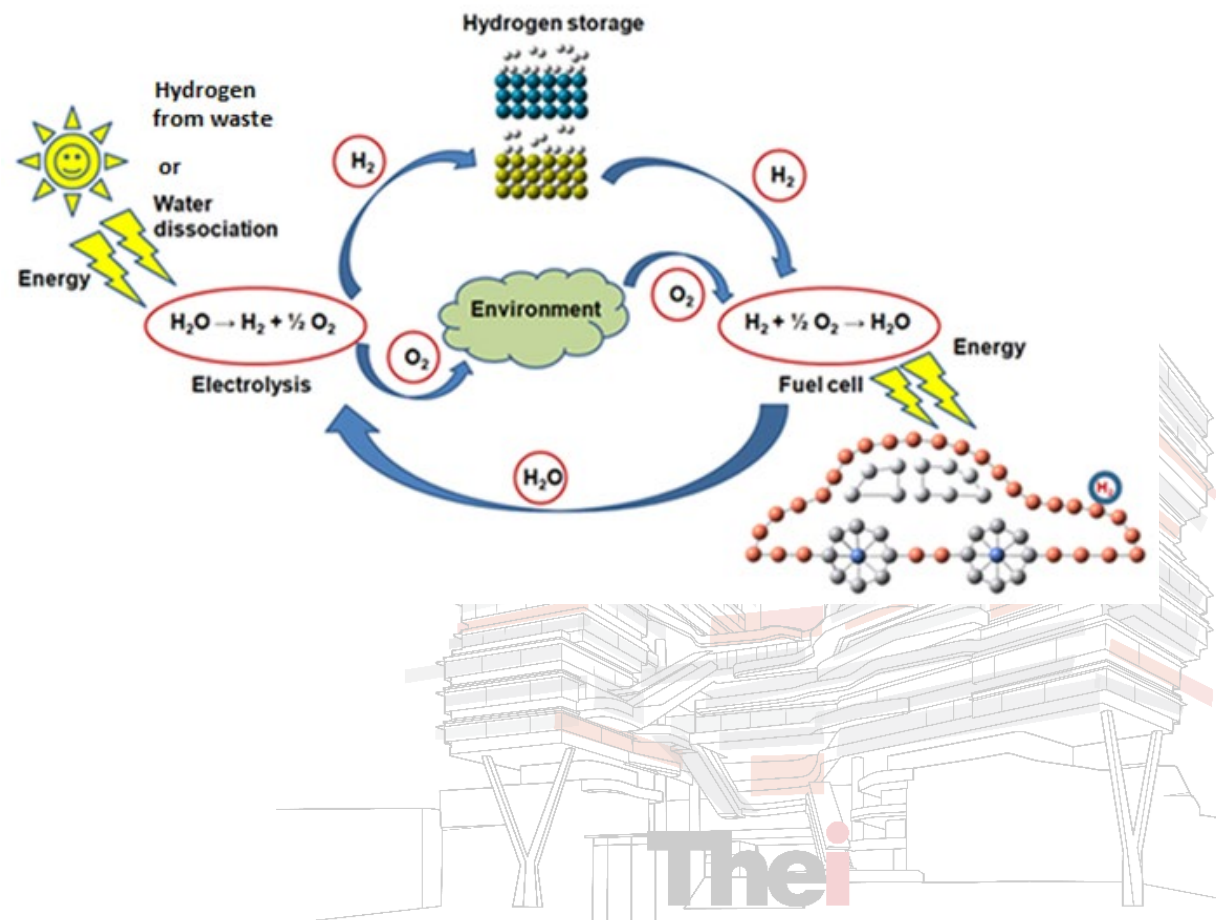
Theoretical Validation





# Conclusion

1. Green H<sub>2</sub> could further accelerate the carbon reduction in Hong Kong for sustainable urban development.
2. Several options for large-scale H<sub>2</sub> storage, such as liquid- and solid-state storage, however, challenges such as kinetics and cycling problem need to be solved.
3. The manufacturing of these HSM could cause another issue in carbon emission, low-carbon chemical production has to be designed.



# Acknowledgement

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UGC/FDS25/E08/20 - Hydrogen on Demand – Development of Hydrogel-Based Hydrogen Generator using Single Atom Strategy for Flexible Power Devices

UGC/FDS25/E04/22 - Development of Synergistic Dual-atom Catalysts with High Activity and Superior Durability for Catalytic Hydrogen Release Reactions

UGC/IDS(R)25/20 - Establishment of the Centre for Interdisciplinary Research on Food By-products Utilization (CIFU)

UGC/IDS(C)14/B(E)01/19 - Development of Renewable Energy for Decarbonizing and Modelling Sustainable Transport and Logistics Operations in Smart Cities of Greater Bay Area

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We also thank for the students and research assistant for the experimental work:

## Students

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## Research Assistant

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# Thank you!

